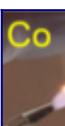
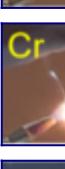
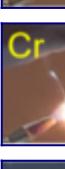
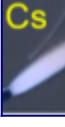


Some common elements and their corresponding colors are:

Symbol	Name	Color[5]	Image
Al	<a href="#">Aluminium</a>	Silver-white, in very high temperatures such as an electric arc, light blue	
As	<a href="#">Arsenic</a>	Blue	
B	<a href="#">Boron</a>	Bright green	
Ba	<a href="#">Barium</a>	Light apple green	
Be	<a href="#">Beryllium</a>	White	
Bi	<a href="#">Bismuth</a>	Azure blue	
C	<a href="#">Carbon</a>	Bright orange	
Ca	<a href="#">Calcium</a>	Brick/orange red; light green as seen through blue glass.	
Cd	<a href="#">Cadmium</a>	Brick red	
Ce	<a href="#">Cerium</a>	Yellow	
Co	<a href="#">Cobalt</a>	Silvery white	
Cr	<a href="#">Chromium</a>	Silvery white	
Cs	<a href="#">Caesium</a>	Blue-violet	
Cu(I)	<a href="#">Copper(I)</a>	Blue-green	

Cu(II)	Copper(II) (non-halide)	Green	
Cu(II)	Copper(II) (halide)	<a href="#">Blue-green</a>	
Fe(II)	<a href="#">Iron(II)</a>	<a href="#">Gold</a> , when very hot such as an electric arc, bright <a href="#">blue</a> , or green turning to orange-brown	
Fe(III)	<a href="#">Iron(III)</a>	Orange-brown	
Ge	<a href="#">Germanium</a>	Pale blue	
H	<a href="#">Hydrogen</a>	Pale blue	
Hf	<a href="#">Hafnium</a>	White	
Hg	<a href="#">Mercury</a>	Red	
In	<a href="#">Indium</a>	<a href="#">Indigo blue</a>	
K	<a href="#">Potassium</a>	Lilac ( <a href="#">pink</a> ); invisible through <a href="#">cobalt blue glass</a> ( <a href="#">purple</a> )	
Li	<a href="#">Lithium</a>	<a href="#">Carmine red</a> ; invisible through <a href="#">green glass</a>	
Mg	<a href="#">Magnesium</a>	Colorless due to Magnesium Oxide layer, but burning Mg metal gives an intense <a href="#">white</a>	
Mn(II)	<a href="#">Manganese(II)</a>	<a href="#">Yellowish green</a>	
Mo	<a href="#">Molybdenum</a>	<a href="#">Yellowish green</a>	
Na	<a href="#">Sodium</a>	Bright yellow; invisible through <a href="#">cobalt blue glass</a> . See also <a href="#">Sodium-vapor lamp</a>	
Nb	<a href="#">Niobium</a>	Green or blue	
Ni	<a href="#">Nickel</a>	Colorless to silver-white	
P	<a href="#">Phosphorus</a>	<a href="#">Pale blue-green</a>	

Pb	<a href="#">Lead</a>	Blue-white	
Ra	<a href="#">Radium</a>	Crimson red	
Rb	<a href="#">Rubidium</a>	Violet red	
S	<a href="#">Sulfur</a>	Blue	
Sb	<a href="#">Antimony</a>	Pale green	
Sc	<a href="#">Scandium</a>	Orange	
Se	<a href="#">Selenium</a>	Azure blue	
Sn	<a href="#">Tin</a>	Blue-white	
Sr	<a href="#">Strontium</a>	Crimson to scarlet red; yellowish through green glass and violet through blue cobalt glass	
Ta	<a href="#">Tantalum</a>	Blue	
Te	<a href="#">Tellurium</a>	Pale green	
Ti	<a href="#">Titanium</a>	Silver-white	
Tl	<a href="#">Thallium</a>	Pure green	
V	<a href="#">Vanadium</a>	Yellowish green	
W	<a href="#">Tungsten</a>	Green	
Y	<a href="#">Yttrium</a>	Carmine, crimson, or scarlet red	
Zn	<a href="#">Zinc</a>	Colorless to blue-green	
Zr	<a href="#">Zirconium</a>	Mild/dull red	